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relationships between the amounts of
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Products. Number of atoms of each
element on the reactant side of the
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Chapter of Class 11 Chemistry for the students so that they can give a quick glance of the chapter. States of Matter. Matter can be defined as anything that occupies volume and has mass. Matter can be classified into three states – solid, liquid and gas. Plasma is regarded as the fourth state

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of matter, which exists only at very high temperatures (at interiors of stars, 10⁷ K). At very high temperatures, all gases ...

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ore. high electrical and thermal
conductivit.... intermediate electrical
and thermal con.... the extraction of
metals from their naturally occurring
materi.... a naturally occurring
material containing a metal. physical
properties of metals. physical
properties of semimetals.

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Measurement. Chemistry is the study
of matter and the changes that
material substances undergo. Of all
the scientific disciplines, it is perhaps

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the most extensively connected to other fields of study. As you begin your study of college chemistry, those of you who do not intend to become professional chemists may well wonder why you need to study chemistry.

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Matter, is perhaps, the most familiar concept of Chemistry that is recapitulated in our Class 11 Chemistry Chapter 1 revision notes. The notes will provide you with a brief overview of the matter and its three

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physical states – Solid – Substances possessing a definite shape and volume are included in this category.

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characteristic that gives a sample of matter its unique identity.

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