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1 Chapter 11 Solution Thermodynamics: Theory In the chemical, petroleum, and pharmaceutical industries multicomponent gases or liquids commonly undergo composition changes as the result of mixing and separation processes, the transfer of species from one phase to another, or chemical reaction.

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2 Chapter 11 Solution Thermodynamics: Theory Because the property of such systems depend strongly on composition as well as on temperature and pressure, our purpose in this chapter is to develop the theoretical foundation for application of thermodynamics to gas mixture and liquid solution. 3 Chapter 11 Solution Thermodynamics: Theory The theory is introduced through derivation of a fundamental property relation for homogeneous solution of variable composition.

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Thermodynamics An Engineering Approach Yunus A. Cengel & Michael A. Boles 7th Edition, McGraw-Hill Companies, ISBN-978-0-07-352932-5, 2008 Sheet 1:Chapter 1 1–5C What is the difference between kg-mass and kg force? Solution Solution

Thermodynamics An Engineering Approach

Chapter 11 Thermodynamics 11.1 Heat, work and energy - Heat is the energy that is the process of being transferred from one to another due to a temperature difference - Work is energy that is being transferred due to the action of a force - James Joule's careful experiments led to the law of conservation of energy - Normally when a substance is heated, its particles gain kinetic energy - During a phase change, the heat energy increases the distance between particles.

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