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~~FuseSchool Ch 2 Matter and  
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Phases of Matter Homogeneous  
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Examples, Classification of  
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~~Pure Substances and Mixtures,  
Elements \u0026amp; Compounds,  
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Chemistry Examples,~~

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~~3.2.1/3.2.2 Describe the  
differences between elements,  
compounds and mixtures. States  
of Matter : Solid Liquid Gas 10  
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Changes in the State of Matter -  
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Liquid, Liquid to Gas, Gas to  
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Chemical Reactions The States of Matter - Changes of State - Solid Liquid Gas Plasma | Physical Science | Gen Chemistry States of Matter Understanding Atoms, elements, and molecules Part #1 (9min) Study Jams - Properties of Matter States of Matter and Changes of State - Science for Kids

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### Change Solutions Manual Chapter

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Chemistry: Matter and Its  
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A change in physical form of a  
substance or physical properties  
of a substance like color, melting  
point, state, size, shape, and  
arrangement of molecules is  
known as physical change. A  
change in chemical properties of  
a substance like composition,  
reactivity with water,  
corrosiveness, and pH is known  
as chemical change. In chemical  
change, a new substance will be  
formed.

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145 CHAPTER 9 SOLUTIONS

MANUAL 35. Aqueous solutions of  
potassium iodide and silver

nitrate are mixed, forming the

precipitate silver iodide. chemical

equation:  $KI(aq) + AgNO_3(aq) \rightarrow$

$KNO_3(aq) + AgI(s)$  complete ionic

equation:  $K^+(aq) + I^-(aq) + Ag^+(aq) +$

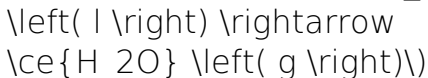
$NO_3^-(aq) \rightarrow K^+(aq) + NO_3^-(aq) +$

$AgI(s)$

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## Solutions Manual Chapter 11 Reactions

Solution. Physical: boiling and melting are physical changes. When water boils, no bonds are broken or formed. The change could be written:



Chemical: The dark grey nail changes color to form an orange flaky substance (the rust); this must be a chemical change.

3.6: Changes in Matter: Physical and Chemical Changes ...

240 Chemistry: Matter and Change □ Chapter 12 Solutions Manual CHAPTER 12 SOLUTIONS MANUAL 30. Compare and contrast sublimation and evaporation. In both processes,

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the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

## States of Matter

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and friendly advice to aid understanding.

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Solutions Manual Chapter

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