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The Lewis concept of acids and bases is discussed in every general, organic and inorganic chemistry textbook. This is usually just a descriptive treatment, as it is not possible to devise a single numerical scale suitable for all occasions. However quantitative Lewis acid-base chemistry can be developed by compiling reaction-specific basicity scales which can be used in specific branches of chemistry and biochemistry.

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order to identify the factors which influence basicity and basicity-dependent properties. In addition, measured experimental basicities and affinities are essential to computational chemists for the validation, calibration and establishment of reliable computational methods for quantifying and explaining intermolecular forces and the chemical bond. Lewis Basicity and Affinity Scales: Data and Measurement is an essential single-source desktop reference for research scientists, engineers, and students in academia, research institutes and industry, in all areas of chemistry from fundamental to applied research. "The book is a noteworthy piece of work and represents a timely and vast accumulation of

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knowledge regarding Lewis bases that brings together accurate thermodynamic and spectroscopic data on typical reference Lewis acids. As such, it should serve as a useful and general guide to basicity." J. AM. CHEM. SOC. 2011, 133, 642

Inorganic Chemistry, Third Edition, emphasizes fundamental principles, including molecular structure, acid-base chemistry, coordination chemistry, ligand field theory and solid state chemistry. The book is organized into five major themes: structure, condensed phases, solution chemistry, main group and coordination compounds, each of which is explored with a balance of topics in theoretical and

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